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Kinetic - molecular theory. Describes the behavior of matter in terms of particles in motion ; Elastic collision. No kinetic energy is lost ; Kinetic energy formula. $KE = \frac{1}{2}mv^2$; Temperature. Measure of the average kinetic energy of the particles in a sample of matter ; Diffusion.

Study Guide - Chapter 12 – States of Matter

Page 1. TEACHER GUIDE AND ANSWERS. Chemistry: Matter and Change. Teacher Guide and Answers. 7. Study Guide - Chapter 12 – States of Matter. Section 12.1 Gases. 1. motion. 2. a. small b. forces c. random d. elastic; kinetic. 3. $KE = \frac{1}{2}mv^2$. 4. Temperature. 5. true. 6. true. 7. false. 8 ...

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Chapter 12: States of Matter

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