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Introduction

Introduction

Solutions

Solubility

Vapour pressure of liquid solutions

Ideal and non-ideal solutions

Colligative properties

Van't Hoff factor

Thank You Bachhon!

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Philosophy of the course

Research facilities

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Stretch your understanding

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What do you expect from the interview

What do you think of your course

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1. MOLECULAR STRUCTURE 2. PRESSURE 3. TEMPERATURE

CRASH COURSE

m (MOLALITY) NUMBER OF MOLES OF SOLUTE PER KILOGRAM OF SOLVENT mol kg PARTIAL PRESSURE

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Internal Energy

Thermochemistry

Infinitesimal Changes

Mathematical Manipulations

Diabatic Changes

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