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What's the empirical formula of calcium phosphate (ions: Ca^{2+} charge and PO_4^{3-} charge)? A. CaPO_4 . B. Ca_2PO_4 . C. Ca_3PO_4 . D. $\text{Ca}_3\text{P}_2\text{O}_8$. D. $\text{Ca}_3\text{P}_2\text{O}_8$.

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a. Write the balanced chemical equation for the reaction that occurs when solutions of barium chloride and sodium carbonate are mixed. Refer to Table 1 on page ...

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Modern Chemistry

SHORT ANSWER Answer the following questions in the space provided. 1. Technological development of a chemical product often.

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Study GuideChapter 5-21 Answer Key

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