Solutions Manual Inorganic 5th Edition Miessler

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Thin layer chromatography - TLC

Limitations of TLC

Gas-liquid chromatography - GC

Retention times

Calibration curve

Limitations of GC

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Introduction

Why do a Titration?

How do you do a Titration?

Precision and Reading the burette.

Concordancy & Mean Titre volume.

Indicators

Concentration calculations

Analysing Titrations

Grid Method for analysis

Grid Method - harder

Percentage Uncertainties in Titrations.

What next?

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focus on the hydrogen oxygen fuel cell

focus on the 1 / 2 equations

so this dotted line here represents the electrolyte

finish by working out the voltage of the cell

come up with the half equations for each electrode

bring the overall charge on the right down to zero

bring the overall charge on the left down to zero

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Core Practical 5 - Hydrolysis of Halogenoalkanes (Topic 10)

Core Practical 6 - Chlorination of 2-methylpropan-2-ol (Topic 10)

Core Practical 6 Past Paper Question

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Physical properties of halogens

Halogen-halide displacement reactions

Colours of the halogens in water and organic solvent

Uses of chlorine

Arguments for and against the chlorination of water

Testing for aqueous halide ions

Summary of halide ion tests

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Carbonate, ...

Intro

Ammonia Ions

Example

Onions

White precipitate

Cream precipitate

lodine precipitate

Correct order

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